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Key indicators

Single-crystal X-ray study
 $T = 150$ K
Mean $\sigma(\text{C}-\text{C}) = 0.007$ Å
 R factor = 0.039
 wR factor = 0.098
Data-to-parameter ratio = 16.2For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

Methylchlorosila[1]ferrocenophane

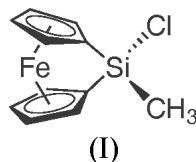
In the title compound, $[\text{Fe}(\text{C}_{11}\text{H}_{11}\text{ClSi})]$, the cyclopentadienyl ligands are tilted towards the bridging Si atom, with a tilt angle of $19.4(3)^\circ$.

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Comment

The title compound, (I), is an important precursor to sila[1]-ferrocenophanes with unsymmetric substitution at the Si atom. The chlorine substituent in (I) can be replaced through reaction with a variety of lithium salts (Jäkle *et al.*, 2000; Berenbaum *et al.*, 2002). When lithium acetylides are used, the $\text{C}\equiv\text{C}$ bond in the functionalized sila[1]ferrocenophanes can further react with dinuclear metal complexes to form highly metallized sila[1]ferrocenophanes with pendant metal clusters (Chan *et al.*, 2003).In (I), the cyclopentadienyl (Cp) ligands are tilted towards the bridging Si atom. The angle between the least-squares planes formed by the Cp ligands bonded to Fe is $19.37(32)^\circ$. The Fe–C distances range from 2.002 (5) to 2.078 (5) Å. The coordination geometry of the Si atom is approximately tetrahedral, with bond angles ranging from $98.0(2)$ to $114.4(2)^\circ$. The tilting of the Cp ligands leads to a compression of the C1–Si1–C6 angle [$98.0(2)^\circ$] compared to other bond angles around Si.In the crystal structure, molecules of (I) related by C -centering are linked into $C(6)$ chains (Bernstein *et al.*, 1995) by weak $\text{C}-\text{H}\cdots\text{Cl}$ interactions (see Table 2 and Fig. 2).

Experimental

Compound (I) was synthesized according to a published procedure (Zechel *et al.*, 1996). Crystals of (I) were obtained by slow evaporation of a solution in hexanes at 298 K under N_2 .

Crystal data

$[\text{Fe}(\text{C}_{11}\text{H}_{11}\text{ClSi})]$
 $M_r = 262.59$
 Monoclinic, Cc
 $a = 13.4146(9)$ Å
 $b = 10.2135(8)$ Å
 $c = 7.8627(6)$ Å
 $\beta = 95.295(5)^\circ$
 $V = 1072.67(14)$ Å³
 $Z = 4$

$D_x = 1.626$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 4024
 reflections
 $\theta = 2.6\text{--}27.5^\circ$
 $\mu = 1.72$ mm⁻¹
 $T = 150(1)$ K
 Block, orange
 $0.20 \times 0.18 \times 0.14$ mm

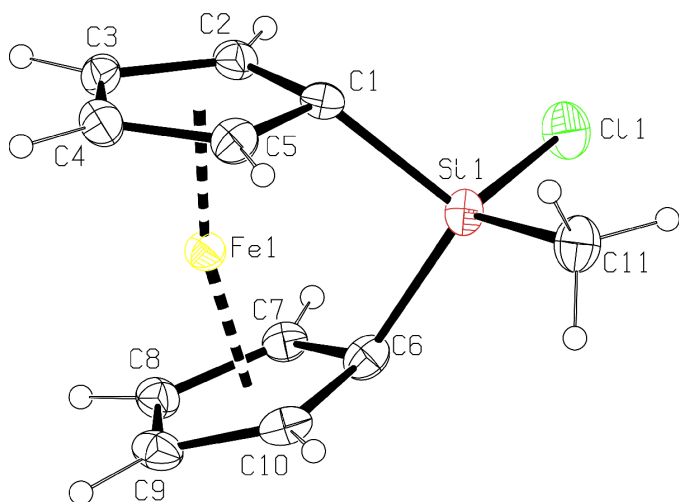


Figure 1
View of (I), with displacement ellipsoids drawn at the 30% probability level (arbitrary spheres for the H atoms).

Data collection

Nonius KappaCCD diffractometer
 φ scans and ω scans with κ offsets
 Absorption correction: multi-scan (SORTAV; Blessing, 1995)
 $T_{\min} = 0.585$, $T_{\max} = 0.783$
 4024 measured reflections
 2078 independent reflections

1859 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.040$
 $\theta_{\text{max}} = 27.5^\circ$
 $h = -17 \rightarrow 15$
 $k = -12 \rightarrow 13$
 $l = -10 \rightarrow 10$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.039$
 $wR(F^2) = 0.098$
 $S = 1.05$
 2078 reflections
 128 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0418P)^2 + 2.256P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.48 \text{ e } \text{Å}^{-3}$
 $\Delta\rho_{\text{min}} = -0.50 \text{ e } \text{Å}^{-3}$
 Absolute structure: Flack (1983),
 845 Friedel pairs
 Flack parameter = $-0.02(2)$

Table 1

Selected geometric parameters (Å, °).

Cl1—Si1	2.0746 (18)	Si1—C1	1.871 (5)
Si1—C11	1.849 (5)	Si1—C6	1.872 (5)
C11—Si1—C1	114.4 (2)	C11—Si1—Cl1	110.5 (2)
C11—Si1—C6	116.0 (3)	C1—Si1—Cl1	107.90 (16)
C1—Si1—C6	98.0 (2)	C6—Si1—Cl1	109.18 (16)

Table 2

Hydrogen-bonding geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
C9—H9A \cdots Cl1 ⁱ	1.00	2.81	3.641 (5)	141

Symmetry code: (i) $x - \frac{1}{2}, \frac{1}{2} + y, z$.

All H atoms were placed in calculated positions, with C—H distances of 0.95 (methyl) and 1.00 Å (Cp), and included in the refinement in the riding-model approximation, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{Cp})$ or $1.5U_{\text{eq}}(\text{methyl})$.

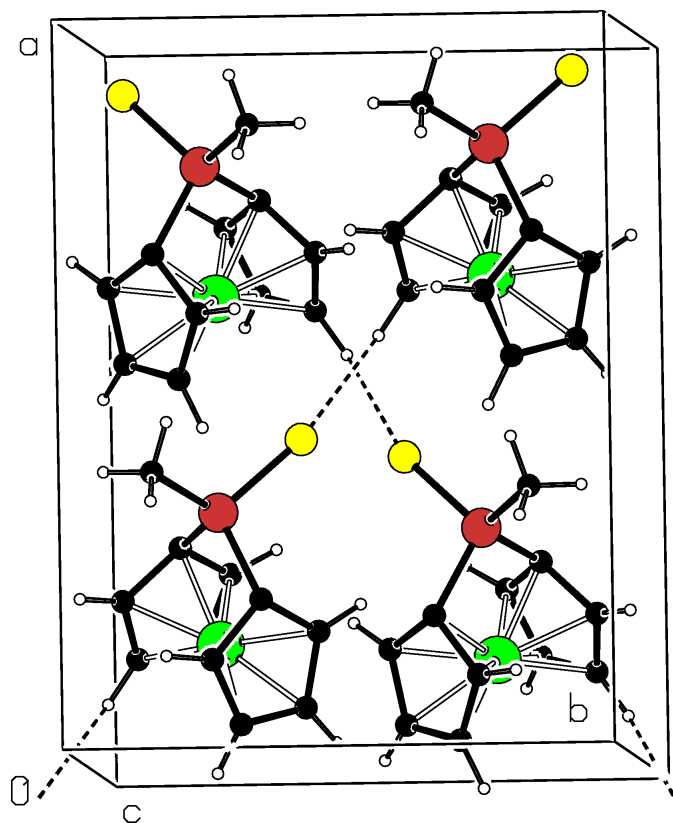


Figure 2

The unit cell contents in (I), viewed approximately along the c axis, showing weak C—H \cdots Cl interactions as dashed lines. Colour codes: green Fe, red Si, yellow Cl and black C.

Data collection: COLLECT (Nonius, 2002); cell refinement: DENZO-SMN (Otwinowski & Minor, 1997); data reduction: DENZO-SMN; program(s) used to solve structure: SHELXTL/PC (Sheldrick, 2001); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL/PC and PLATON (Spek, 2003); software used to prepare material for publication: SHELXTL/PC.

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